Tetrahedron Letters No. 19, 1613 - 1614, 1976. Pergamon Press. Printed in Great Britain.

CAGE COMPOUNDS. 3. ¹³C AND ¹H N.M.R. TO CHARACTERISE SYMMETRICAL REARRANGEMENT PRODUCTS OF 3,6-EPOXY PENTACYCLO[6,2,2,0^{2,7},0^{4,10},0^{5,9}]DODECANE WITH SULPHUERIC ACID.

J. Dekker*, J.J. Dekker, L. Fourie

Department of Chemistry, Potchefstroom University, Potchefstroom 2520, Republic of South Africa.

T.G. Dekker

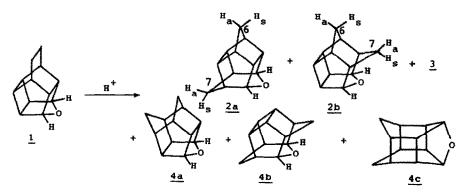
Department of Pharmaceutics, Potchefstroom University.

K.G.R. Pachler and P.L. Wessels

Chemical Physics Section, National Chemical Research Laboratory, C.S.I.R., P.O. Box 375, Pretoria 0001, Republic of South Africa.

(Received in UK 16 March 1976; accepted for publication 29 March 1976)

Various skeletal rearrangements of strained bird-cage compounds to less strained isomers have been reported in recent years. We now wish to report that the title compound $(\underline{1})^2$ rearranges on treatment with 95% sulphuric acid (neat) at room temperature for 24 hrs. Three isomeric ethers $(\underline{2}, \underline{3} \text{ and } \underline{4})$, which were separated by gas chromatography , were formed in the ratio 27:3:1. The IR spectra were free from carbonyl and alcohol absorptions. The absence of olefinic bonds in 2, $\underline{3}$ and $\underline{4}$ was shown by $\overline{^{1}}$ H and $\overline{^{13}}$ C nmr.



The proton noise-decoupled ¹³C nmr spectrum of 2 displayed 7 signals [CDCl₃: 82.6(2C), 53.9(2C), 52.1(2C), 44.9(1C), 44.4(2C), 41.0(2C) and 29.2(1C) ppm downfield from TMS]. In the undecoupled spectrum the signals at 44.9 and 29.2 ppm were split respectively into a triplet (J = 131.1 Hz) and a doublet of doublets (J = 130.6 and 134.0 Hz), thus representing two non-identical methylene carbons. The remaining signals were essentially split into doublets. These results point to a symmetrical structure (2a or 2b). The ¹³C LIS experiments [Yb(fod)₃] indicated the methylene carbon at 29.2 ppm to be closer to the oxygen atom than the methylene carbon at 44.9 ppm. The same result was obtained from the ¹H LIS experiments [Eu(fod)₃], en=abling us to identify the methylene protons H_{7s}, H_{7a}, H_{6s} and H_{6a}. Since no NOE effect on H_{6s} or H_{6a} was observed upon irradiation of H_{7a}, one is inclined to favour structure 2a rather than 2b for this isomer. This seems reasonable, since 2a is expected to be less strained than 2b.

The proton noise-decoupled 13 C nmr spectrum of $\underline{4}$ exhibited only four signals [CDCl $_3$: 81.9(4C), 55.2(4C), 40.5(4C) and 32.0(2C) ppm downfield from TMS] indicating a highly symmetri= cal structure. The undecoupled spectrum of $\underline{4}$ showed that the peak at 32.0 ppm corresponds to two identical methylene carbons. The 1 H nmr spectrum of $\underline{4}$ displayed two different pairs of methylene protons at 1.26 and 1.76 ppm ($J_{\text{gem}} = 11.7 \text{ Hz}$). Since the 1 H LIS [Eu(fod) $_3$] of these two signals are small and correspond to those of protons H_{6s} and H_{6a} of $\underline{2}$, little doubt exists that $\underline{4}$ should have structure $\underline{4a}$ rather than $\underline{4b}$. The alternative symmetrical structure 4c is highly improbable due to its expected high internal strain.

Compound 3 is totally dissymmetric, as shown by its 13 C proton noise-decoupled spectrum [CDCl₃: 85.5, 77.2, 56.2, 50.4, 48.6, 47.4, 47.0, 46.6, 45.8, 42.1, 40.1 and 27.9 ppm downfield from TMS]. From this data it is impossible, at this stage, to assign a structure to 3.

We are presently studying the details of this novel rearrangement.

Acknowledgement

We are indebted to the C.S.I.R. for financial support.

References

- (1) a. A.P. Marchant, T-C. Chou and M. Barfield, <u>Tetrahedron Lett.</u>, 3359 (1975) and references cited therein.
 - b. L.A. Paquette, R.S. Beckley and T. McCreadie, <u>Tetrahedron Lett.</u>, 775 (1971) and references cited therein.
- J. Dekker, J.J. Dekker, L. Fourie and G.L. Wenteler, J.S. African Chem. Inst., 28, 321 (1975).
- 3. 6m x 9mm column, 10% PEG20M on Chromosorb P 60~80 mesh (treated with 10% KOH), 180°, F.I.D.
- 4. R. Freeman, H.D.W. Hill, B.L. Tomlinson and L.D. Hall, J. Chem. Phys., 61, 4466 (1974).